# SEM6\_C13\_ORGANOMETALLIC \_CATALYSIS

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## SEM6\_C13\_ORGANOMETALLIC\_CATALYSIS

(1+3)=

2

#### MATHABHANGA COLLEGE INTERNAL ASSESMENT EXAM, 2023

Answer the following questions:

1. (a) Draw the structure of Wilkinson's catalyst. Describe its catalytic activity towards hydrogenation of alkene.

(b) What are the disadvantages in hydroformylation reaction using  $Co_2(CO)_8$  as catalyst. Give the catalytic cycle of that reaction. (1+3)=4

(c) Why is the Ir based catalyst catalyst most efficient than the Rh based catalyst.

2. (a)  $[PdCl_4]^{2-}$  is not suitable catalyst for oxidation of alkene to aldehyde by Wacker process-explain. Write down the catalytic cycle of that process. (2+3)=5

(b) Write an example of catalyst for polymerization of alkene. How it can be prepared? Why it is more efficient than Zeiglar catalyst. (1+2+2)=5

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Answer the following questions:

3. (a) Draw the structure of Wilkinson's catalyst. Describe its catalytic activity towards	
hydrogenation of alkene.	(1+3)=4
(b) What are the disadvantages in hydroformylation reaction using Co <sub>2</sub> (CO) <sub>8</sub> as catalyst.	Give the catalytic cycle
of that reaction.	(1+3)=4
(c) Why is the Ir based catalyst catalyst most efficient than the Rh based catalyst.	2

4. (a)  $[PdCl_4]^{2-}$  is not suitable catalyst for oxidation of alkene to aldehyde by Wacker process-explain. Write down the catalytic cycle of that process. (2+3)=5

(b) Write an example of catalyst for polymerization of alkene. How it can be prepared? Why it is more efficient than Zeiglar catalyst. (1+2+2)=5

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Answer the following questions:

5. (a) Draw the structure of Wilkinson's catalyst. Describe its catalytic activity towards hydrogenation of alkene. (1+3)=4
(b) What are the disadvantages in hydroformylation reaction using Co<sub>2</sub>(CO)<sub>8</sub> as catalyst. Give the catalytic cycle of that reaction. (1+3)=4
(c) Why is the Ir based catalyst catalyst most efficient than the Rh based catalyst. 2

6. (a) [PdCl<sub>4</sub>]<sup>2-</sup> is not suitable catalyst for oxidation of alkene to aldehyde by Wacker process-explain. Write down the catalytic cycle of that process. (2+3)=5

(b) Write an example of catalyst for polymerization of alkene. How it can be prepared? Why it is more efficient than Zeiglar catalyst. (1+2+2)=5